**Programme: M. Sc. Part-I (Chemistry)** 

Course Code: OCO-401

Title of the Course: Synthetic Organic Chemistry I

Number of Credits: 03 Effective from AY: 2018-19

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|--------------------|---|-------|
| Prerequisites      | Should have studied the courses / topics in Organic Chemistry at F Y B Sc,  |       |
| for the course:    | S Y B Sc and T Y B Sc levels as well as the course CHOC-401 so as to  |       |
|                    | have basic knowledge of organic nomenclature and basic principles.  |       |
| Course             | 1. Introduction to concepts of functional groups and their interconversion  |       |
| <b>Objectives:</b> | 2. Learning mechanistic concepts of carbon-carbon bond making by  |       |
| <b>g</b>           | nucleophilic addition to carbonyl group   |       |
|                    | 3. Learning mechanistic aspects of various oxidation & reductionprocesses   |       |
|                    |   |       |
|                    | used in organic syntheses.  |       |
| Course             | 1. Students should be in a position to choose appropriate oxidizing agent   |       |
| Outcomes:          | for oxidation of a particular functional group.   |       |
|                    | 2. Students should be in a position to choose appropriate reducing agent for  |       |
|                    | reduction of a particular functional group.   |       |
|                    | 3. Students shall be in a position to understand/propose plausible  |       |
|                    | mechanism of organic reactions.   |       |
|                    | 4. Student should be able to choose appropriate nucleophilic addition   |       |
|                    | reaction for making carbon-carbon bond.   |       |
|                    | reaction for making curbon curbon bond.   |       |
| Content:           | 1. Oxidation reactions:   | 11    |
| Content:           |   |       |
|                    | Oxidation of organic compounds using chromium (PCC, PDC) and  | hrs   |
|                    | manganese compounds, Oppenauer oxidation, Swern oxidation,  |       |
|                    | ozonolysis. Other methods of oxidation such as selenium dioxide,  |       |
|                    | Pb(OAc) <sub>4</sub> , HIO <sub>4</sub> , peracids, peroxides, OsO <sub>4</sub> ,RuO <sub>4</sub> , DMSO (Swern) sodium |       |
|                    | bromated / CAN & NaOCl, DDQ, Prevost's reagent and Woodward   |       |
|                    | Conditions;   |       |
|                    | Catalytic oxidation over Pt, Photosensitised oxidation of alkenes, oxidation  |       |
|                    | with molecular oxygen, aromatization, silver based reagents.  |       |
|                    | 2.Reduction reactions:  | 9 hrs |
|                    | Reduction of organic compounds using hydride-transfer reagents and  |       |
|                    | related reactions: MPV reduction, NaBH <sub>4</sub> , Trialkylborohydrides, LAH   |       |
|                    | & lithium hydridoalkoxyaluminates, mixed LAH-AlCl <sub>3</sub> reagents,  |       |
|                    | DIBAL and reduction with borane and dialkylboranes, Enzymatic   |       |
|                    | reduction involving liver alcohol dehydrogenase/NADH & Bakers' yeast,   |       |
|                    | catalytic hydrogenation, Dissolving metal reductions including acyloin  |       |
|                    | condensation, Clemmensen reduction and Birch reduction, Other methods   |       |
|                    |   |       |
|                    | of reduction: Wolff-Kishner, Raney Ni desulphurisation, di-imide.   |       |
|                    | 3.Halogenation:   | 5 hrs |
|                    |   | JIIIS |
|                    | Formation of Carbon Halogen bonds: Substitution in saturated compounds,   |       |
|                    | alcohols, carbonyl compounds, substitution at allylic and benzylic  |       |
|                    | compounds, bromodecarboxylation (Hunsdiecker reaction), Finkelstein   |       |
|                    | reaction, iodolactonisation.  |       |

|             | 4. <b>Esterifiction, amide preparation and hydrolysis:</b> (study of different mechanisms and reagents)   | 6 hrs |
|-------------|---|-------|
|             | 5. Name reactions: Knoevenegel Reaction, Claisen, Darzen, Stobbe, Perkin, Aldol, Benzoin, Pechmann condensation.                                      | 5 hrs |
| Pedagogy:   | Mainly Lectures & tutorials. Seminars / assignments / presentations / self-study or a combination of some of these could also be used to some extent. |       |
| References/ | 1. H. O. House, <i>Modern Synthetic Reactions</i> , 2 <sup>nd</sup> Ed., W. A. Benjamin,  |       |
| Readings    | Benjamin-Cummings Publishing Co., 1972.   |       |
|             | 2. W. Caruthers, <i>Modern Methods of Organic Synthesis</i> , 4 <sup>th</sup> Ed.,Cambridge University Press, 2004.                                   |       |
|             | 3. M. B. Smith, Jerry March, <i>Advanced Organic Chemistry- Reaction</i> , <i>Mechanism and Structure</i> , 6 Ed, Wiley, 2006.                        |       |
|             | 4. F.A. Carey & R.J. Sundberg, <i>Advanced Organic Chemistry</i> (Part A & B) 5 <sup>th</sup> Ed., Springer India Private Limited, 2007.              |       |
|             | 5. P Sykes, <i>A guidebook to mechanisms in organic chemistry</i> , 6 <sup>th</sup> Ed., Pearson Edu., 1996.  |       |
|             | 6. Clayden, Greeves, Warren and Wothers, <i>Organic Chemistry</i> , 2 <sup>nd</sup> Ed., Oxford University Press, 2002.                               |       |
|             | 7. E.S. Gould, <i>Mechanism and structure in Organic Chemistry</i> , Holt, Reinhart and Winston 1965.   |       |
|             | 8. F. A. Carey, R. M. Giuliano, <i>Organic Chemistry</i> , 8 <sup>th</sup> Ed., McGraw-Hill, 2010.  |       |
|             | 9. S.H. Pine, <i>Organic Chemistry</i> , 5th Ed, McGraw-Hill International Edn. McGraw-Hill, 1980.  |       |