**Programme: M. Sc. Part-I (Chemistry)** 

**Course Code: ICO-401** 

Title of the Course: Topics in Inorganic Chemistry & Environmental Chemistry

Number of Credits: 03 Effective from AY: 2018-19

Prerequisites for the	Student should have studied the courses in chemistry at F.Y. B.Sc., S.Y.B.Sc. and T.Y.BSc. levels and / or CHIC-401 course so as to have basic knowledge	No. of lectures
course:	of Inorganic / environmental chemistry.	lectures
Course Objectives:	<ol> <li>To provide fundamental aspects of transition &amp; inner transition metals &amp; their compounds.</li> <li>To provide knowledge of main group elements of the periodic table &amp; their compounds</li> <li>To introduce various global phenomenon's of atmosphere &amp; environment, follow directive of the Supreme Court in 1993 to introduced environmental education at all levels, have a fair knowledge on the various global activities to justify permissible or adverse, so that future generation are not adversely affected.</li> </ol>	
Course Outcomes:	<ol> <li>Students should be in position to understand fundamentals / usefulness of transition &amp; inner transition metals.</li> <li>Students should be in position to understand chemistry main group elements.</li> <li>Students shall be aware of the maintenance of healthy living atmosphere on the globe.</li> </ol>	
Content:	SECTION-I	
	1. Chemistry of transition & inner transition elements 1.1 Transition elements: IUPAC definition of transition elements, occurrence, physical & chemical properties, noble character, metal oxides & oxido complexes, examples of metal-metal bonded clusters.	9 hr
	1.2 Inner transition elements: Lanthanides, occurrence, properties, oxidation states, electronic structure, colour and spectra, magnetic properties, lanthanide contraction, compounds of lanthanides. Actinoid chemistry, general trends.	
	<ul> <li>2. Main group elements and their compounds</li> <li>2.1 Boron group: Compounds of boron:-borazine and boron nitride, synthesis, properties, structure &amp; bonding. Borates: classification, structures &amp; examples.</li> </ul>	9 hr
	2.2 Carbon group: Allotropes of carbon including C <sub>60</sub> , intercalation compounds of graphite, carbides. Compounds of silicon: silicates, zeolites & silicones.	
	2.3 Nitrogen group:- Introduction: oxides & oxyacids of nitrogen. 2.4 Oxygen group: oxyacids & oxohalides of S, S <sub>4</sub> N <sub>4</sub> ring compounds: synthesis, properties, structure & bonding.	

	SECTION-II	
	1. <b>Atmosphere</b> Structure and properties of the atmosphere, composition of atmosphere and vertical temperature behaviour, lapse rate and temperature inversion.	2 hr
	2. <b>Air Pollution</b> Classification of air pollutants and photochemical reactions in the atmosphere Common air pollutants (e.g. CO, NOx, SO <sub>2</sub> , hydrocarbons and particulates) (a) sources (b) physiological and environmental effect (c) monitoring, d) various remedial & technological measures to curb pollution. Air quality standards.	7 hr
	3. Water pollution Importance of buffer & buffer index in waste water treatments. C hemical, physical & biological characteristics of water pollution, specific & non-specific characterization of water. DO, BOD, COD, and chlorine demand, typical water treatment & waste water treatment (Municipal).	5 hr
	4. Treatment of Industrial wastes Electroplating industry, fertilizer industry and pharmaceuticals industries.	2 hr
	5. Biogeochemical cycles: Carbon and Nitrogen cycles nature	2 hr
Pedagogy:	Mainly lectures / tutorials. Seminars / assignments / presentations / self-study or a combination of some of these could also be used to some extent.	
Text books / reference books	<ol> <li>P.W. Atkins, T. Overton, J. Rourke, M. Weller, &amp; F. Armstrong, Shriver &amp; Atkins Inorganic Chemistry, Oxford publications, 2009, 5<sup>th</sup> Ed.</li> <li>J. E. Huheey, E. A. Kieter, R. L. Kieter &amp; O. K. Medhi, Inorganic Chemistry: Principles of Structure &amp; Reactivity, Pearson, 2011, 4<sup>th</sup> Ed.</li> <li>F. A. Cotton, G. Wilkinson &amp; P. L. Gauss, Basic Inorganic Chemistry, Wiley, 2008 (reprint), 3<sup>rd</sup> Ed.</li> <li>N.N. Greenwood and A. Earnshaw, Chemistry of the Elements, Pergamon Press, Exetr, Great Britain. 1984.</li> <li>J.D. Lee, Concise Inorganic Chemistry, Wiley, 2008, 5<sup>th</sup> Ed.</li> <li>A.V. Salker, Environmental Chemistry: Pollution and Remedial Perspective, Narosa Publication, 2017.</li> <li>A.K. De, Environmental Chemistry, New Age, 2006.</li> <li>A.C. Stern, R.W. Boubel, Fundamentals of Air Pollution, D. Bruce turner &amp; D.L.Fox, Academic Press, 1984.</li> <li>R.A. Horne, Chemistry of Our Environment", John Wiley, N.Y. (1978).</li> <li>C.N. Sawyer &amp; P.J. Macarty, Chemistry for Environmental Engineering, Mc Graw Hill, 1978.</li> <li>L.L. Ciaccio, Water and Water Pollution Hand Book", Marcel Dekker, 1973.</li> <li>J.C. Lamb, Water Quality and its Control, John Wiley &amp; Sons, N.Y., 1985.</li> </ol>	